

## Effect of Mild Steel Particles on the Thermal, Frictional and Wear Characteristics of Ceramic Matrix Composites

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### Abstract:

Micro particles of mild steel, silicon carbide, magnesia and bentonite were employed as input materials in the production of ceramic matrix composites by powder metallurgy. The composites were characterised and they demonstrated desirable characteristics. Sample D containing 12 wt. % of mild steel particles exhibited a very good thermal stability in the temperature range of 0 – 800 °C, high coefficient of friction of 0.59 and reduced wear rate of  $2.228 \times 10^{-5}$  g/m which implies high resistance to wear. Proper blending of constituent materials and strong bonding enhanced the characteristics of the composites. The results indicated the suitability of the composite for use in fields that require high resistance to thermal decomposition and wear.

**Keywords:** Ceramic matrix composites, Micro particles, Thermal decomposition, Friction, Wear rate.

### 1. INTRODUCTION

The need for the development of friction materials that will exhibit superlative characteristics especially high resistance to thermal decomposition and wear cannot be over emphasized. This is because of the immense benefits, which such materials will offer. Ceramics are suitable materials for wear resistance components because of their hardness, corrosion resistance, and high thermal stability, which have made them to be extensively used for tribological applications [1]. Generally, the performance of materials under different operating conditions is influenced by the composition, structure, interfaces, grain size, and properties of constituents such as hardness, mechanical strength, wear resistance, and thermal conductivity [2, 3].

The high hardness and abrasion resistance of silicon carbide (SiC) have made carbon fiber/SiC composites the choice materials for frictional purposes. They exhibit superlative characteristics such as high coefficient of friction, high resistance to wear and corrosion, heat resistance as they can function very well at temperatures up to 1200 °C, low density, low coefficient of linear thermal expansion, and the absence of noise during

braking [4]. Because of these desirable characteristics, they are widely used in the brake systems of automobiles and aircrafts. Materials made of ceramic composites which are used in various applications experience mechanical abrasion and thermal wear in service. Wear and coefficient of friction represent interface instability that have effect on the life span of friction materials under contact and repeated cyclic motion. They could be indirect measure of contact surface degradation that affects materials quality [3]. Friction is the resistance to movement when a solid surface moves over another surface. The two surfaces that are rubbing against each other experience deformation/degradation [1]. For example, brake pads must have a sufficiently high coefficient of friction with the brake disc to achieve braking effectiveness and should not decompose at high temperature so that the coefficient of friction with the brake disc is not adversely affected/reduced [5, 6].

Generally, wear control is challenging due to some factors such as environment (temperature, humidity, oxygen content), contact (normal load, sliding speed, type of motion), and materials properties such as hardness, surface roughness, contaminations, porosity and interfacial bonding of constituents [7]. However,

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ceramic matrix composites were developed to solve the problems of wear and thermal decomposition. Ceramic matrix composites dispersed with metallic particles are among the materials that have potentials for high functional applications under severe environment such as high temperature. These composites offer the possibility of combining resistance to heat, wear and corrosion because of the ceramic phase with mechanical strength offered by the presence of metallic phase [8]. The oxidation behaviour of Zirconia matrix composite dispersed with stainless steel particles was investigated. The result indicated that the ceramic composite exhibited a different corrosion behaviour at 700 °C and 800 °C compared to pure stainless steel because of the presence of Zirconia [8]. Carbon fiber reinforced silicon carbide matrix composites were developed by liquid-phase siliconizing technology for energy intensive brake system application. The results indicated that the

composites demonstrated a relatively high coefficient of friction and minimal wear rate [4]. Hence, this study seeks to investigate the effect of mild steel particles on the thermal, frictional and wear characteristics of ceramic matrix composites for possible application in areas that require high heat and wear resistance.

## 2. MATERIALS AND METHODS

### 2.1 Materials and Preparation

Silicon carbide, mild steel particles, bentonite and magnesia are the input materials. Mild steel chips obtained from the Engineering workshop of the University of Lagos were ground and sieved to the desired particles size using British standardised sieves (BSS). As-received particles of silicon carbide, magnesia and bentonite were obtained in Lagos. Figure 1 shows the materials used.



**Figure 1.** Input materials used for the study (a) mild steel chips (b) 105- $\mu\text{m}$  ground and sieved mild steel particles (c) As-received 53- $\mu\text{m}$  silicon carbide (d) As-received 90- $\mu\text{m}$  bentonite (e) As-received 90- $\mu\text{m}$  magnesia.

### 2.2 Production of the Composites

105- $\mu\text{m}$  mild steel particles were blended with 53- $\mu\text{m}$  silicon carbide, 90- $\mu\text{m}$  magnesia and 90- $\mu\text{m}$  bentonite. Varied formulations were used in producing the samples as illustrated in Table 1. Phenolic resin and water were mixed with each of the formulations. The mixtures were fed into metal moulds rubbed with oil and pressed at 0.35

MPa to enhance their surface smoothness. The samples were removed from the moulds and sun-dried for 12 hrs. They were oven dried at 110 °C for 12 hrs to avoid cracking during sintering. They were sintered at 1100 °C using a muffle furnace and soaked for 1 hr after which they were taken out of the furnace and allowed to cool. Few samples are presented (Figure 2).

**Table 1.** Proportion of used materials (wt. %).

Sample	Mild steel particles	Silicon carbide	Magnesia	Bentonite	Phenolic resin	Total
A	0	30	40	25	5	100
B	4	35	31	25	5	100
C	8	40	22	25	5	100
D	12	45	13	25	5	100
E	16	50	4	25	5	100



**Figure 2.** Photograph of some of the ceramic composite samples produced.



**Figure 3.** Experimental set-up of the flame resistance test on the composites.

### 2.3 Characterisation of the Samples

A scanning electron microscope was employed to reveal the microstructure of the samples. To test for the flame resistance of the composites, each of the five samples was weighed before the test and their initial weights were recorded. They were put on wire gauze placed directly on the flame of a Bunsen burner for one hour. The flame was removed and the samples were left to cool and their final weights were recorded. The weight loss obtained is presented in Table 3. The experimental set-up for this test is shown in Figure 3.

Thermogravimetric method was used in determining the thermal stability of the samples with the aid of a thermogravimetric analyser. The analyser measured the weight of the samples as they were heated at varied temperatures. The onset temperature of destruction ( $T_{des}$ ) of the samples, which indicates their decomposition temperature was recorded. The coefficient of friction of the samples was determined by making them slide over a smooth mild steel plane inclined at  $35^\circ$  that had a pulley with loads applied on the hanger. The coefficient of friction ( $\mu$ ) between each sample and the steel surface was determined by employing Equations 1 and 2 [9].

$$\mu = \frac{\text{Sliding force}}{\text{Normal force}} \quad 1$$

$$\mu = \frac{P - W \sin \theta}{W \cos \theta} \quad 2$$

Where:

P = Load on the hanger (N)

W = Weight of sample (N)  
 $\theta$  = Angle of inclination

Dry sliding abrasive wear test was conducted by employing a pin-on-disc wear tester according to ASTM standard G99. The disc rotated with the aid of a D.C. motor of speed 250 rpm and sliding occurred between the stationary samples that had contact with the rotating disc. The wear rate ( $\omega$ ) was calculated using Equations 3 – 5 [10].

$$\text{Wear rate } (\omega) = \frac{\Delta w}{S} \quad 3$$

Where:

### 3. RESULTS AND DISCUSSION

#### 3.1 Microstructure

The microstructures show that the samples contain different phases with different shapes. Some are globular while some are needle-like as revealed in the micrographs. The elemental composition as revealed by

$$\Delta w = w_0 - w_1 \quad 4$$

$$S = \text{sliding speed} = 2\pi N D t = 1256.8 \text{ m} \quad 5$$

$\Delta w$  = weight loss in g

$w_0$  = initial weight of sample before wear in g

$w_1$  = final weight of sample after wear in g

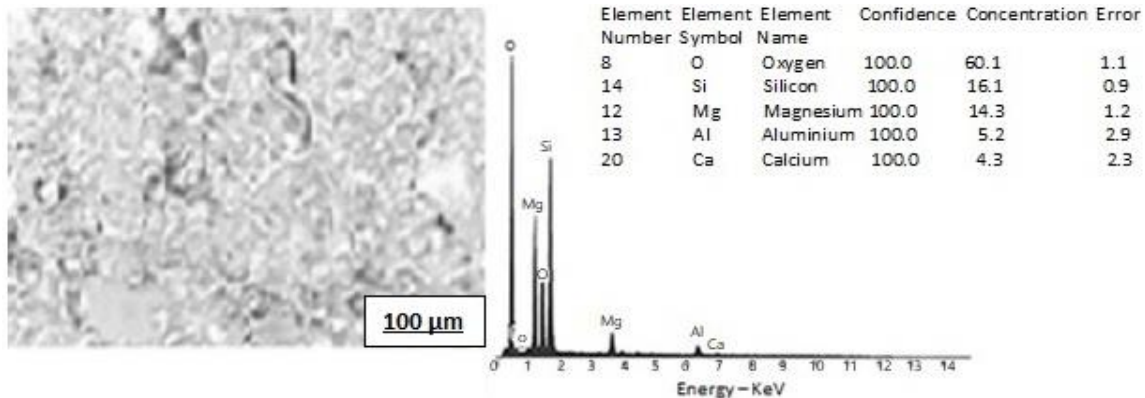
D = diameter of the grit sandpaper which is 0.16 m

N = number of revolution per minute (rpm) of the pin-on disc which is 250

t = exposure time to wear = 5 mins

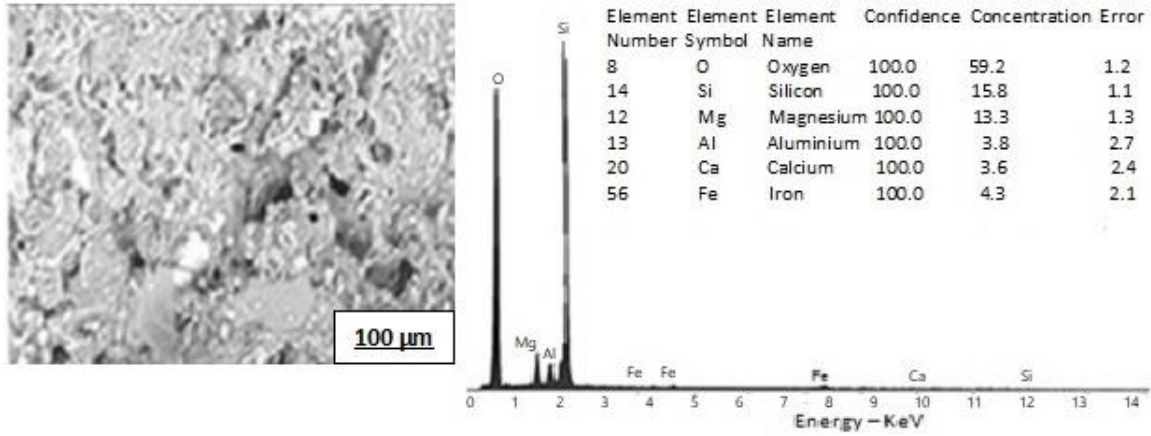
$\pi = 3.142$ .

the energy-dispersive spectroscopy (EDS) spectra show that sample A contains O, Si, Al, Mg and Ca. Other samples also contain these elements plus Fe from added iron mild steel particles shown in Figure 4 – 8. The whitish area in the micrographs is magnesia (MgO) phase while the dark area is silicon carbide phase and gray is a combination of Fe and bentonite phase.

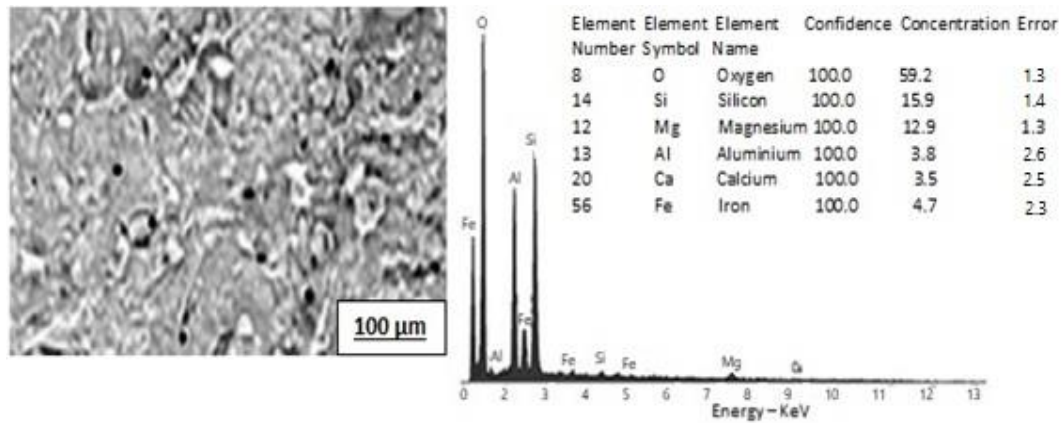


**Figure 4.** Scanning electron micrograph (SEM) of sample A with the EDS spectrum showing the phases and the elemental composition.

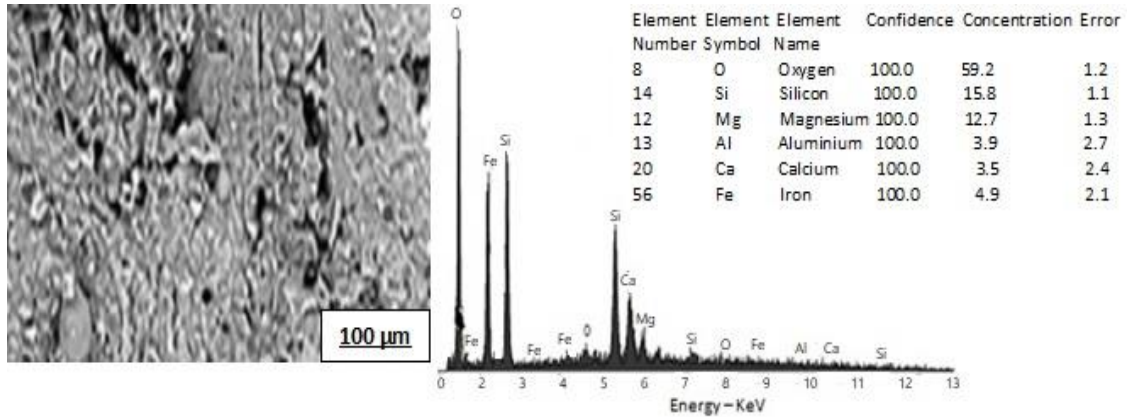




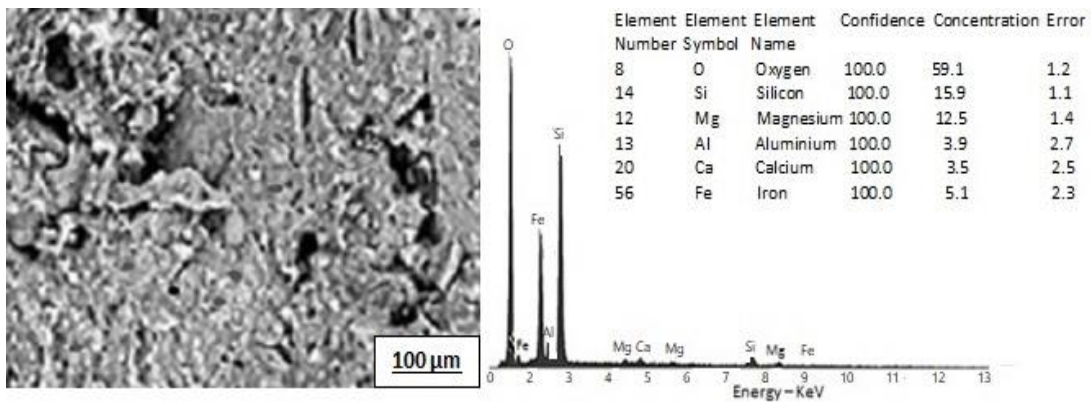
**Figure 5.** Scanning electron micrograph (SEM) of sample B with the EDS spectrum showing the phases and the elemental composition.



**Figure 6.** Scanning electron micrograph (SEM) of sample C with the EDS spectrum showing the phases and the elemental composition.



**Figure 7.** Scanning electron micrograph (SEM) of sample D with the EDS spectrum showing the phases and the elemental composition.



**Figure 8.** Scanning electron micrograph (SEM) of sample E with the EDS spectrum showing the phases and the elemental composition.

### 3.2 Heat Resistance of the Samples

After 1 hr of exposing (Figure 3) the samples to heat, the initial weight of the samples before exposure to heat was the same as the final weight after exposure except sample A. Sample A lost 0.01 g which is 0.015 %. This shows that 99.9 % of the weight of samples A was retained but

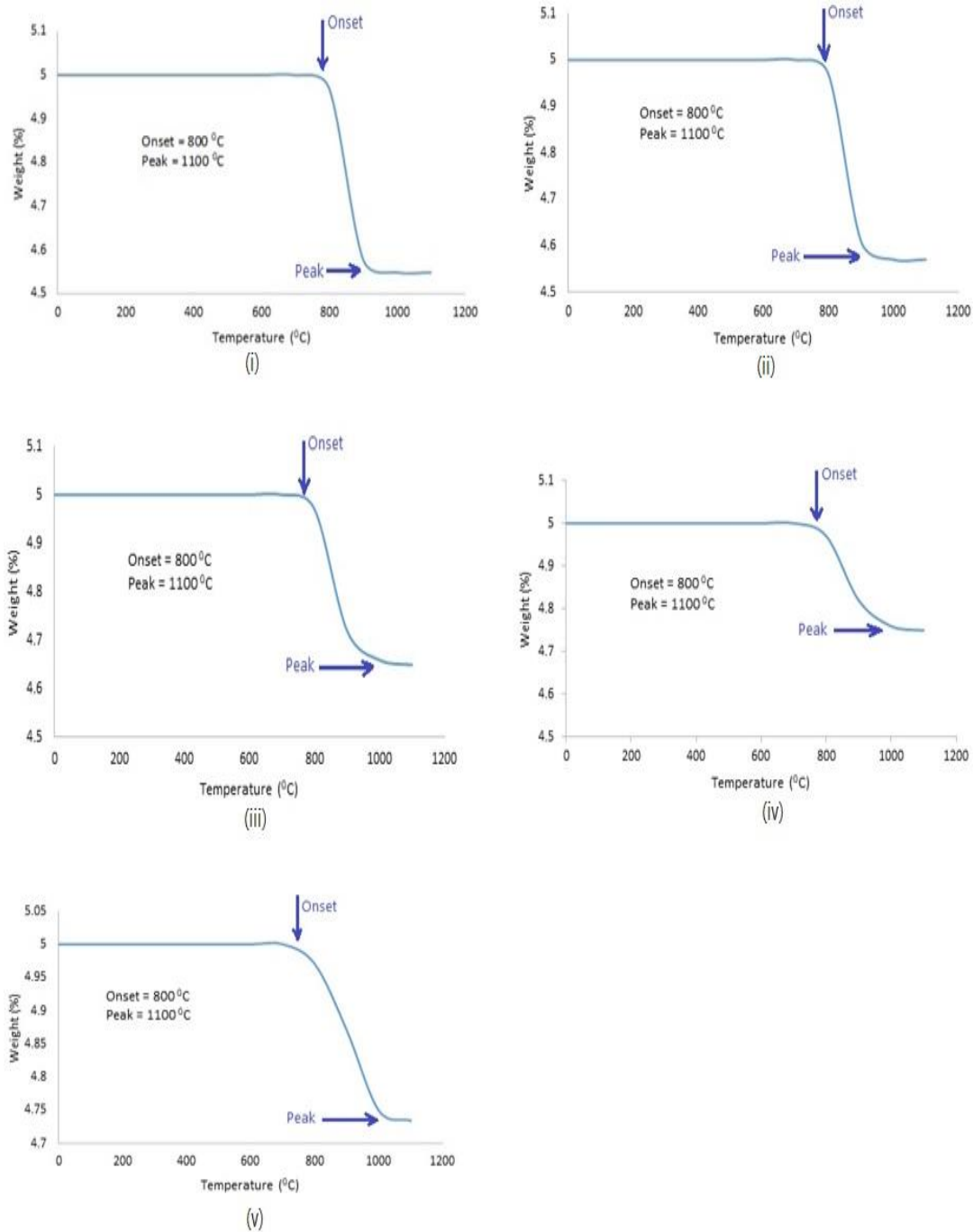
100 % of the weight of samples B, C, D and E was retained indicating that there was no weight loss in them. It implies that the samples demonstrated excellent resistance to thermal decomposition, which may be because of the refractory nature of ceramics and the strong bond existing between the particles.

**Table 2.** Effect of heat on the samples.

Sample	Description	Initial Weight (g)	Final Weight (g)	Weight loss (g)	Weight loss (%)	Weight Retained (%)
A	Not mixed with mild steel particles	67.17	67.18	0.01	0.015	99.985
B	Mixed with 4 wt. % mild steel particles	156.51	156.51	-	-	100
C	Mixed with 8 wt. % mild steel particles	158.09	158.09	-	-	100
D	Mixed with 12 wt. % mild steel particles	159.75	159.75	-	-	100
E	Mixed with 16 wt. % mild steel particles	162.23	162.23	-	-	100

Furthermore, results of the thermogravimetric test show that the weight loss due to thermal decomposition is minimal as illustrated in Figure 9. Sample A demonstrated 0.45 % loss in weight, which is higher than that of others. Specifically, sample D demonstrated the least loss in weight of 0.24 %. Thermal decomposition started from 800 °C (onset) to the peak (1100 °C) with reduction in weight of the samples. The weight loss of

the samples is in the range of 0.24 – 0.45 %. This is very low and it indicates that the composites exhibited high resistance to thermal decomposition, which may be because of the refractory nature of ceramics and the strong bond existing between the particles. The strong bond existing between the particles served as a shield against thermal decomposition, which agrees with the report by [11].



**Figure 9.** Thermogravimetric plot of the samples showing the decomposition temperatures (onset and peak) and the weight (i) sample A, (ii) sample B, (iii) sample C, (iv) sample D, (v) sample E.



### 3.3 Coefficient of Friction

As shown in Figure 10, there is a progressive increase in the coefficient of friction ( $\mu$ ) of the samples. This shows that asperities exist between the surfaces of the mild steel and the samples. The rise in coefficient of friction is

promoted by the rising weight of samples. The range of the coefficient of friction which is 0.52 – 0.59, is more than the specification range (0.3 – 0.38) for automobiles brake linings/pads by the Nigerian industrial standard [12] which is a positive development.

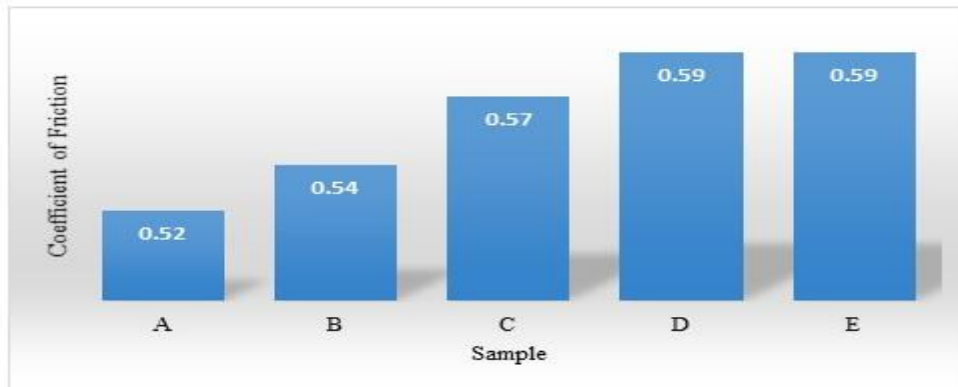


Figure 10. Coefficient of friction of the samples.

### 3.4 Loss in Weight and Wear Rate of the samples

The samples experienced loss in weight because of surface deformation due to ploughing force of asperities as presented in Figure 11. A similar submission was reported by [13]. The same trend is seen in Figure 12

since weight loss and wear rate are directly related. Samples D and E exhibited the lowest wear rate ( $2.228 \times 10^{-5} \text{ g/m}$ ). The strong bond existing between the particles reduced the possibility of particles pullout and led to the reduced wear rate of the samples.

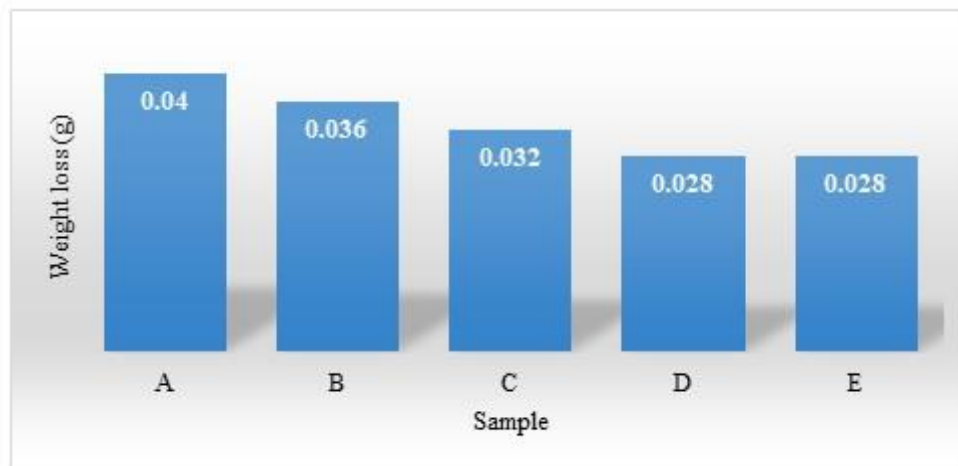


Figure 11. Loss in weight due to wear of the samples.

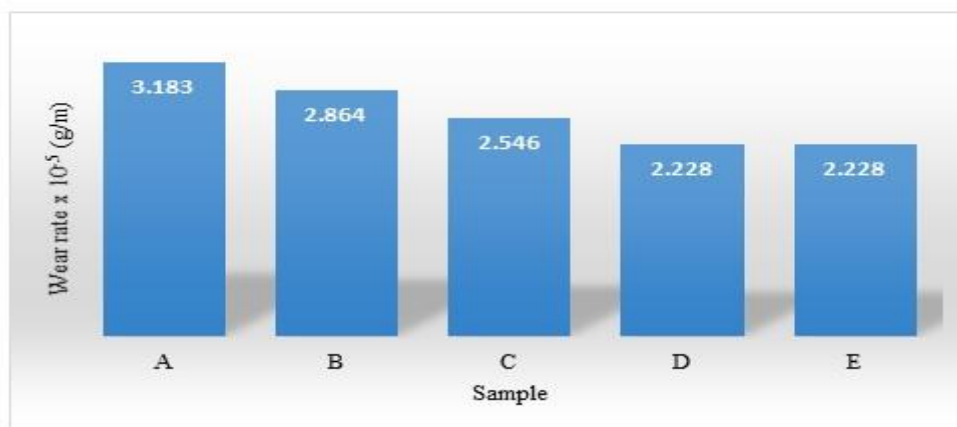


Figure 12. Wear rate of the samples.

#### 4. CONCLUSION

Particulate ceramic matrix composites were produced and characterised. The composites demonstrated very good properties. Sample D did not experience any weight loss when heated for 1 hr and exhibited no decomposition in the temperature range of 0 – 800 °C. It also exhibited a high coefficient of friction of 0.59 and minimal wear rate of  $2.228 \times 10^{-5}$  g/m. These indicate that the sample demonstrated high thermal stability/heat resistance and wear resistance.

#### Conflict of Interests

The authors declare no conflict of interest.

#### Declaration of Ethical Standards

The authors of this article declare that the materials and methods used in this study do not require ethical committee permission and/or legal-special permission.

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