



## EFFICIENT AND SELECTIVE ANTITUMOR AGENTS BASED ON CATIONIC CALIXARENES: SYNTHESIS, CHARACTERIZATION, AND ANTIPROLIFERATIVE PROPERTIES

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**ABSTRACT:** Research work on the preparation of anti-tumor agents with active properties is still a major field for the pharmaceutical industry to promote more potent anticancer drugs to the market. However, one of the main disadvantages of current and future anticancer drugs is selectivity. Therefore, in recent decades, a new perspective has to be developed for chemotherapy, such as "targeted" drugs and minimal side effects. Calixarenes, composed of phenolic units linked by methylene bridges are versatile kinds of macrocyclic compounds in supramolecular chemistry that can be modified to hydrophilic and hydrophobic cavities. The biocompatibility of calixarene derivatives allows them to be used in the development of in vivo and in vitro applications. For this reason, the calixarenes with different active groups, have been synthesized by many researchers as a target structure, and their biological activities and in particular their anti-cancer properties, have been studied. The purpose of the current study is to synthesize calix[4]arene derivatives bearing the cationic group at the lower rim and investigation their cytotoxic effect for several cancerous cells. Results demonstrated that calix[4]arene derivative having 5-bromopentyl trimethylammonium bromide moieties (CN) and calix[4]arene derivative having 3-bromopropyl-triphenylphosphonium bromide moieties (CP) compounds selectively inhibits proliferation of A549 (13.42  $\mu\text{M}$ ) and HeLa (17.05  $\mu\text{M}$ ) and Hep-2 cells (>200  $\mu\text{M}$  and 162.71  $\mu\text{M}$ ), respectively.

**Keywords:** Supramolecule, Calixarene, Cytotoxicity, Cationic, Anticancer Agent

### Katyonik Kaliksarenlere Dayalı Etkili ve Seçici Antitümör Ajanlar: Sentez, Karakterizasyon ve Antiproliferatif Özellikler

**ÖZ:** Aktif özelliklere sahip anti-tümör ajanların hazırlanmasına yönelik araştırma çalışmaları, farmasötik endüstrisinin piyasaya daha güçlü antikanser ilaçları tanıtması için hala önemli bir alandır. Bununla birlikte, mevcut ve gelecekteki antikanser ilaçlarının ana dezavantajlarından biri seçiciliktir. Bu nedenle, son yıllarda kemoterapi için "hedeflenen" ilaçlar ve minimal yan etkiler gibi yeni bir bakış açısı geliştirilmelidir. Metilen köprüleriyle bağlı fenolik birimlerden oluşan kaliksarenler, supramoleküler kimyada hidrofilik ve hidrofobik boşluklara dönüştürülebilen çok yönlü makrosiklik bileşiklerdir. Kaliksaren türevlerinin biyouyumluluğu, in vivo ve in vitro uygulamaların geliştirilmesinde kullanılmalarına izin verir. Bu nedenle farklı aktif gruplara sahip kaliksarenler birçok araştırmacı tarafından hedef yapı olarak sentezlenmiş ve biyolojik aktiviteleri ve özellikle kanser önleyici özellikleri incelenmiştir. Bu çalışmanın amacı, alt kenarda katyonik grubu taşıyan kaliks[4]aren türevlerini sentezlemek ve çeşitli kanserli hücreler için sitotoksik etkilerini araştırmaktır. Sonuçlar, CN ve CP bileşiklerinin sırasıyla A549 (13.42  $\mu\text{M}$ ) ve HeLa (17.05  $\mu\text{M}$ ) ve Hep-2 hücrelerinin (>200  $\mu\text{M}$  ve 162.71  $\mu\text{M}$ ) proliferasyonunu seçici olarak engellediğini gösterdi.

**Anahtar Kelimeler:** Supramolekül, Kaliksaren, Sitotoksikite, Katyonik, Antikanser Ajan

## 1. INTRODUCTION

Cancer, i.e., uncontrolled growth of abnormal cells, is the second leading cause of death in the world after cardiovascular diseases (Siegel et al., 2019). Pharmaceutical industries throughout the world produce many anticancer agents, most of them have negative side effects, and they cannot distinguish selectively tumor cells from healthy cells. To overcome this situation, efforts are made by different scientists and they are working mainly on hybrid molecules containing two or more pharmacophores' groups that can selectively target tumor cells without effecting healthy cells (Geraci et al., 2008; Pur, 2016). Moreover, supramolecular chemistry has provided a more sophisticated solution in the form of macrocyclic compounds such as calixarenes. Calix[n]arene are third-generation supramolecular class with eclectic applications after cyclodextrin and crown ethers. They are synthesized by base catalyzed condensation of para-substituted phenol and formaldehyde. They consist of central annulus phenolic rings connected by methylene units, at the lower part hydroxyl groups and a hydrophobic cavity on the upper rim (Akceylan et al., 2021; Cengel ve Farabi, 2021; Consoli et al., 2015; Oguz et al., 2020; Oguz et al., 2020; Oguz et al., 2020; Oguz et al., 2021; Ozcelik, Farabi ve Tabakci, 2019). Calixarenes have excellent flexible structural properties that can be modified according to its application. Recently, calixarenes have been proven to be good building scaffolds for the synthesis of anticancer agents in the literature (Rouge et al., 2010; Nimse ve Kim, 2013; Santos et al., 2015; Yousaf et al., 2015; Naseer et al., 2017; Yilmaz et al., 2020; An et al., 2021; Zhou et al., 2021). Due to their multiple derivatization possibilities, different functional groups that have anti-tumor properties can easily be incorporated at their lower/upper rim (de Fátima et al., 2009; Rodik et al., 2009; Dings et al., 2013; Saluja ve Sekhon, 2013; Läppchen et al., 2015; Yousaf et al., 2015; An et al., 2016; An et al., 2019; Rego ve ark., 2019).

Here, based on all of the above considerations, we have focused on synthesizing two cationic group-bearing calix[4]arene derivatives to investigate their anticancer activity in different human cancer cell lines and human healthy epithelial cell line.

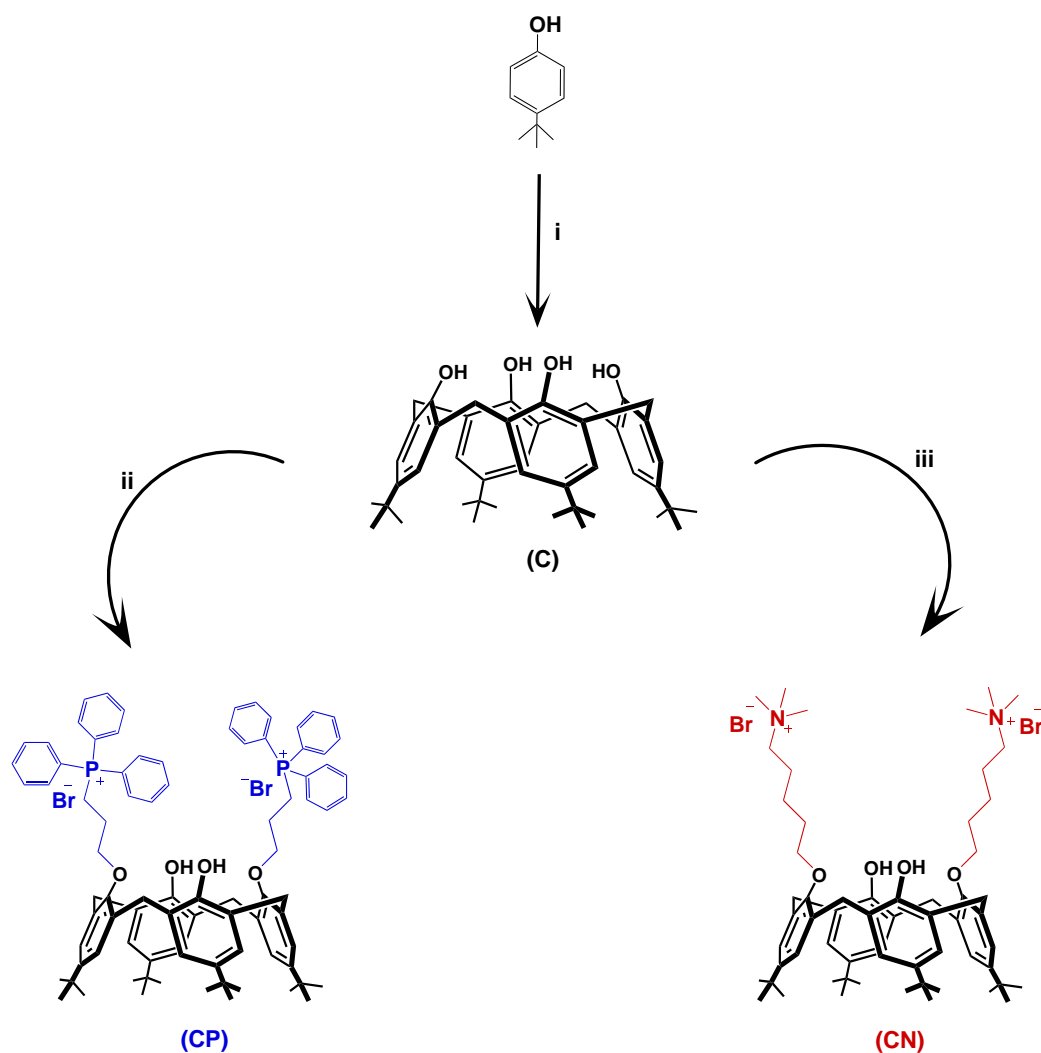
## 1. EXPERIMENTAL SECTION

### 1.1. Methods, Materials and Instruments

All chemicals and solvents used in the experimental studies were supplied as the analytical grade from various commercial companies like Merck and Sigma-Aldrich. Infrared spectra (FT-IR) measurements were performed using a Vertex 70 (ATR) instrument spectrometer. Varian (400 MHz) instrument was used for NMR measurements ( $^1\text{H}$  and  $^{13}\text{C}$  NMR) of both **CN** and **CP** in deuterated solvents.

### 1.2. Synthesis of Compound CP

The compound **CP** has been synthesized according to the literature with some modifications (Sayin ve Yilmaz, 2017). A mixture of *p*-tert-butylcalix[4]arene (**C**) (0.75 g, 1.15 mmol) and (3-bromopropyl)triphenylphosphonium bromide (1.21 g, 3.25 mmol) in acetone (30 mL) was refluxed in the presence of  $\text{K}_2\text{CO}_3$  (0.27 g, 2.87 mmol) for 36 h. The reaction was checked with TLC until compound **C** disappeared. The excess of the solvent was evaporated in a vacuum and the remaining product was extracted with chloroform/ $\text{H}_2\text{O}$ . The chloroform phase was dried with anhydrous magnesium sulfate. The target product (**CP**) was obtained as a white solid (yield: 88%).  $^1\text{H}$ -NMR (400 MHz,  $\text{CDCl}_3$ ): 1.08 (s, 18H, *t*-butyl), 1.54 (s, 18H, *t*-butyl), 2.53 (s, 4H, - $\text{CH}_2$ ), 3.32 (d, 4H,  $J = 13.2$  Hz, Ar- $\text{CH}_2$ -Ar), 4.22 (d, 4H,  $J = 13.3$  Hz, Ar- $\text{CH}_2$ -Ar), 4.47-4.47 (m, 8H, - $\text{CH}_2$ ), 6.77 (s, 2H, ArOH), 6.84 (s, 4H, ArH), 7.52 (s, 4H, ArH), 7.91-8.14 (m, 30H, ArH).  $^{13}\text{C}$ -NMR (100 MHz,  $\text{CDCl}_3$ ): 17.5, 22.2, 31.2, 31.9, 33.7, 34.1, 34.3, 75.0, 118.1, 119.0, 125.9, 127.6, 130.8, 132.9, 134.0, 135.4, 135.6, 142.1, 147.5, 149.1, 150.3.



**Figure 1.** The synthesis pathway for the preparation of CP and CN. Reaction conditions: (i) HCHO, NaOH; (ii)  $K_2CO_3$ , acetone, 3-bromopropyl-triphenyl-phosphonium bromide; (iii)  $K_2CO_3$ , acetone, 5-bromopentyl-trimethylammonium bromide.

### 1.3. Synthesis of Compound CN

The compound CN was synthesized with some modifications according to the previous method (Sayin ve Yilmaz, 2017). In acetone (30 mL), *p*-tert-butylcalix[4]arene (C) (0.75 g, 1.56 mmol), (5-bromopentyl)trimethylammonium bromide (0.94 g, 3.27 mmol) and  $K_2CO_3$  (0.54 g, 3.92 mmol) were refluxed for 20 h. The reaction was checked with TLC until compound C disappeared. The excess solvent was evaporated in a vacuum and the remaining product was extracted several times with chloroform/ $H_2O$ . The chloroform phase was dried with anhydrous magnesium sulfate. The target product (CN) was obtained as a white solid (yield: 90%).  $^1H$ -NMR (400 MHz,  $CDCl_3$ ): 1.17 (s, 18H, *t*-butyl), 1.52 (s, 18H, *t*-butyl), 2.02 (bs, 4H, -CH<sub>2</sub>), 2.27-2.36 (bs, 8H, -CH<sub>2</sub>), 3.56 (d, 4H,  $J = 13.3$  Hz, Ar-CH<sub>2</sub>-Ar), 3.69-3.73 (s, 18H, N-CH<sub>3</sub>), 3.99 (bs, 4H, -CH<sub>2</sub>), 4.22 (bs, 4H, -CH<sub>2</sub>), 4.44 (d, 4H,  $J = 13.3$  Hz, Ar-CH<sub>2</sub>-Ar), 7.12 (s, 4H, ArH), 7.29 (s, 4H, ArH), 7.39 (s, 2H, ArOH).  $^{13}C$ -NMR (100 MHz,  $CDCl_3$ ): 22.3, 22.8, 29.2, 31.2, 31.3, 36.5, 31.8, 34.0, 34.4, 52.7, 65.8, 76.2, 125.6, 126.0, 128.1, 133.5, 141.9, 147.4, 150.3, 150.5.

#### 1.4. Cells and Cell Culture

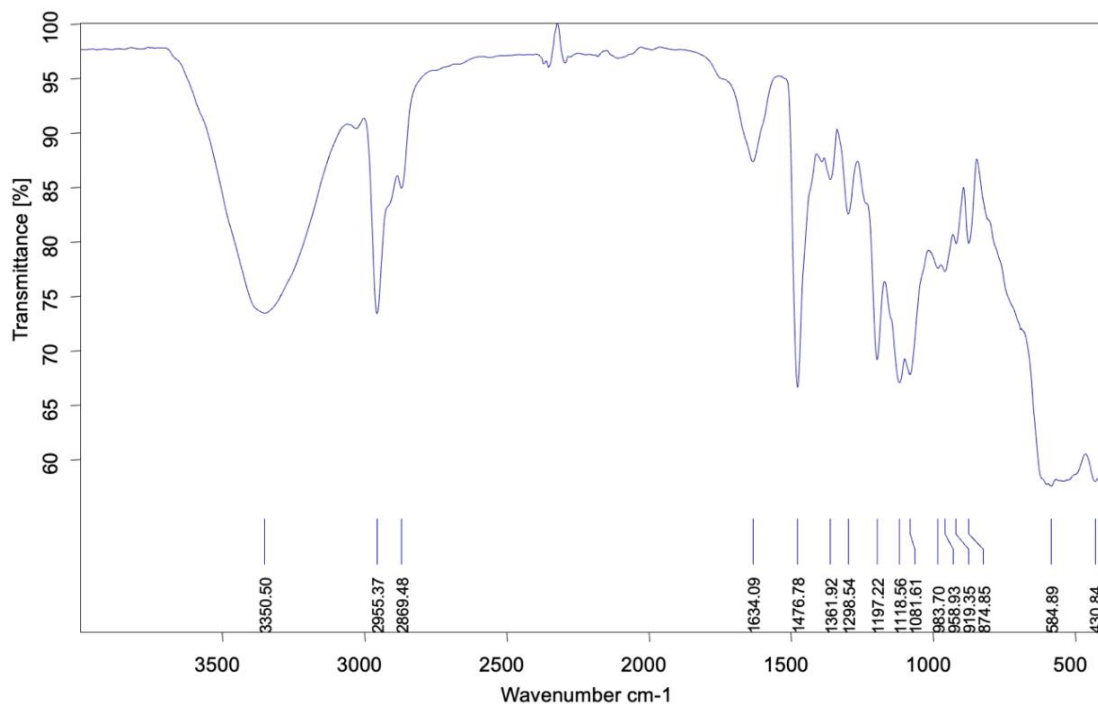
A-549 (human Lung cancer line), HeLa (human cervix cancer line) and healthy epithelial cell line, Hep-2 were obtained from ATCC (American Type Culture Collection). Cells were incubated with DMEM F-12K, EMEM and MEM mediums, respectively and supplemented with 10% FBS (fetal bovine serum), antibiotics (100 U/mL penicillin and 100 µg/mL streptomycin) and 2 mM L-glutamine at 37 °C in a 5% CO<sub>2</sub> atmosphere and 95% humidity.

#### 1.5. Alamar Blue Cell Viability Assay

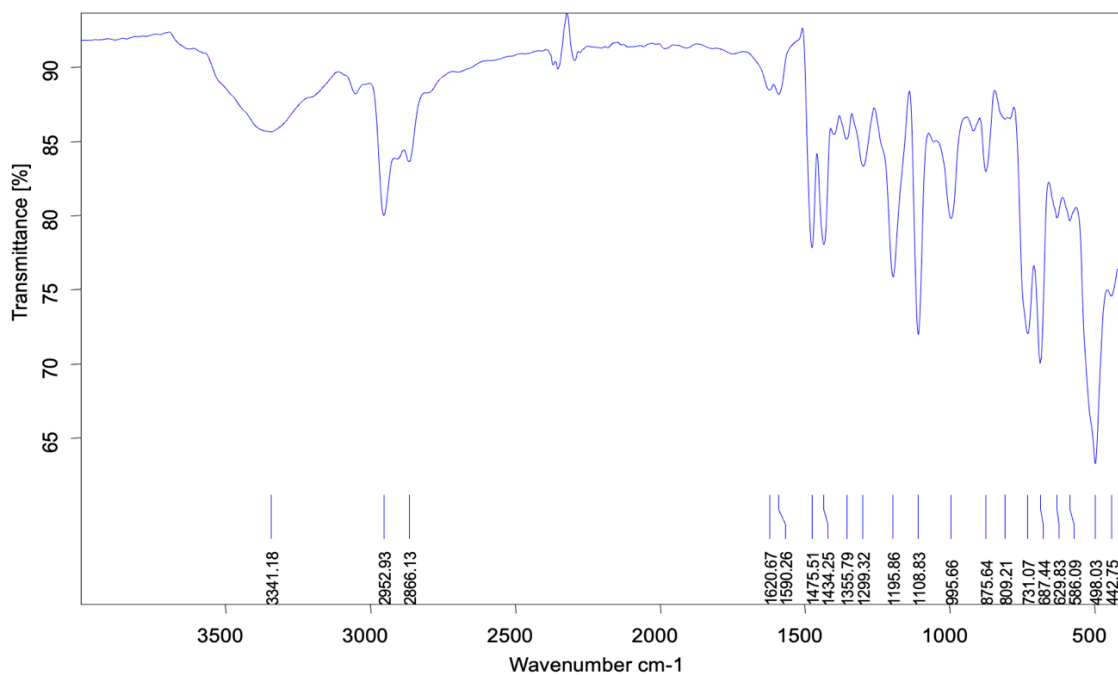
Cell viability was determined by the Alamar Blue method (Invitrogen, Thermo Fischer Scientific, Waltham, MA, USA). Cells in the sub-culture stage were harvested from the flask by treatment with trypsin [0.05% in PBS (pH 7.4) containing 0.02% EDTA (Erdemir ve ark., 2021)]. The IC<sub>50</sub> value of compounds CN and CP were detected from the sigmoidal graph of cell inhibition, statistical analyzes were performed with GraphPad Prism software.

## 2. RESULTS AND DISCUSSION

The intent of current study was to synthesize antiproliferative agents from calixarene by containing cationic groups like 3-bromopropyl-triphenyl-phosphonium bromide and 5-bromopentyl-trimethylammonium bromide (**Figure 1**). As depicted in Figure 1, Compound CN and CP were synthesized by reacting *p*-tert-butylcalix[4]arene with 3-bromopropyl-triphenyl-phosphonium bromide and 5-bromopentyl-trimethylammonium bromide in acetone e for 20 h and 36 h, respectively. The formation of CN was approved by the presence of specific bands at 1197 cm<sup>-1</sup> for C-N, 3350 cm<sup>-1</sup> for Ar-OH in the FTIR spectra (Fig. 2). The formation of CP was confirmed by the appearance of the characteristic Ar-P bands at about 1475 cm<sup>-1</sup> and Ar-OH at 3341 cm<sup>-1</sup> in the FTIR spectra (Fig. 3).

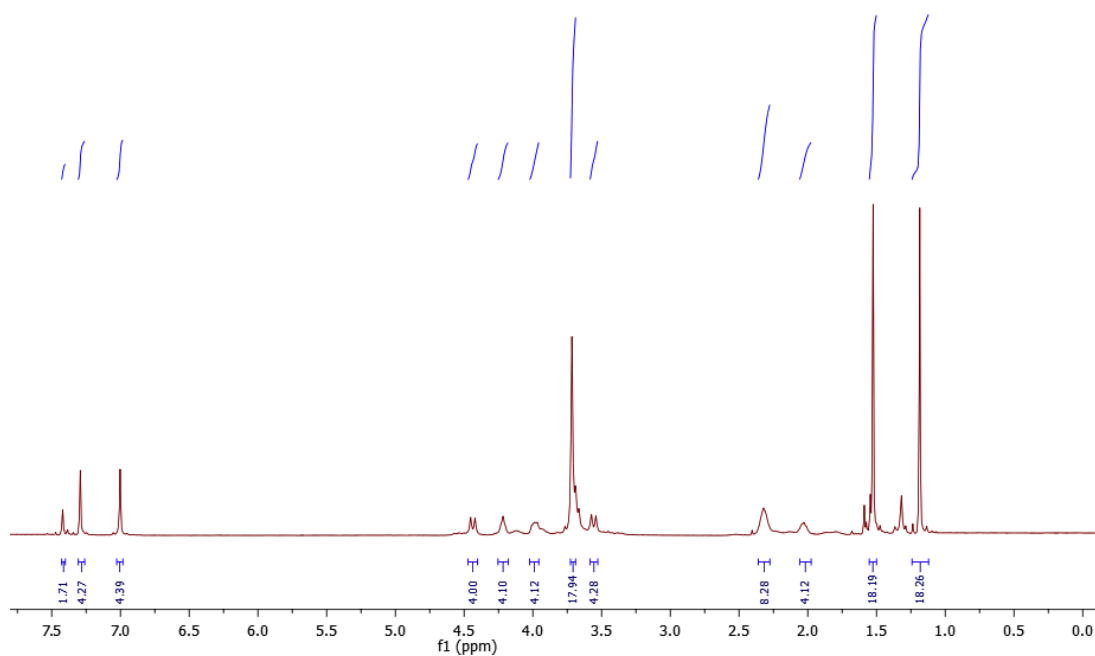


**Figure 2.** FTIR spectrum of compound CN



**Figure 3.** FTIR spectrum of compound CP

According to <sup>1</sup>H NMR spectra, the structure of CN was approved by the presence of -CH<sub>2</sub> signals between 2.22 and 4.44 ppm, the existence of two p-tert butyl signals (-C(CH<sub>3</sub>)<sub>3</sub>) at 1.17 ppm and 1.52 ppm. In addition, the presence of the N-CH<sub>3</sub> signals at 3.75 ppm, ArH signals at 7.01 ppm and 7.29 ppm proved the successful synthesis of CN (Fig. 4).



**Figure 4.** <sup>1</sup>H NMR spectrum of compound CN

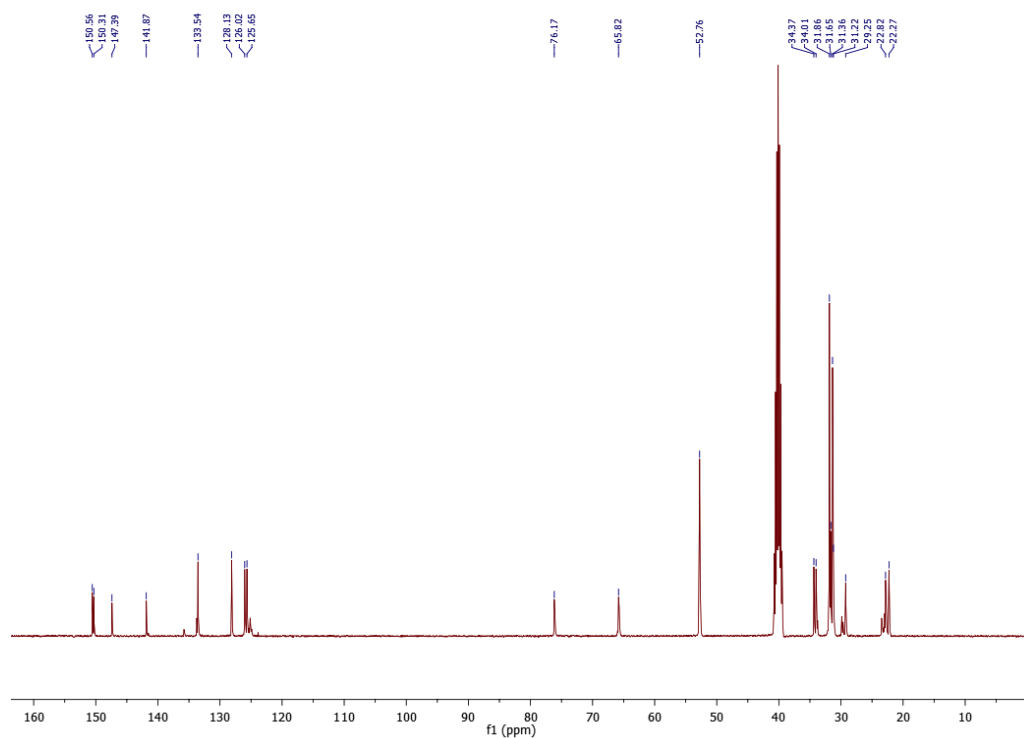


Figure 5. <sup>13</sup>C NMR spectrum of compound CN

The existence of the peaks at 1.08 ppm and 1.54 ppm for p-tert-butyl groups (-C(CH<sub>3</sub>)<sub>3</sub>), the peaks between 2.53 ppm and 4.47 ppm for -CH<sub>2</sub> groups, and the presence of new ArH protons from 7.91 to 8.14 ppm was confirmed the successful synthesis of CP (Fig. 6).

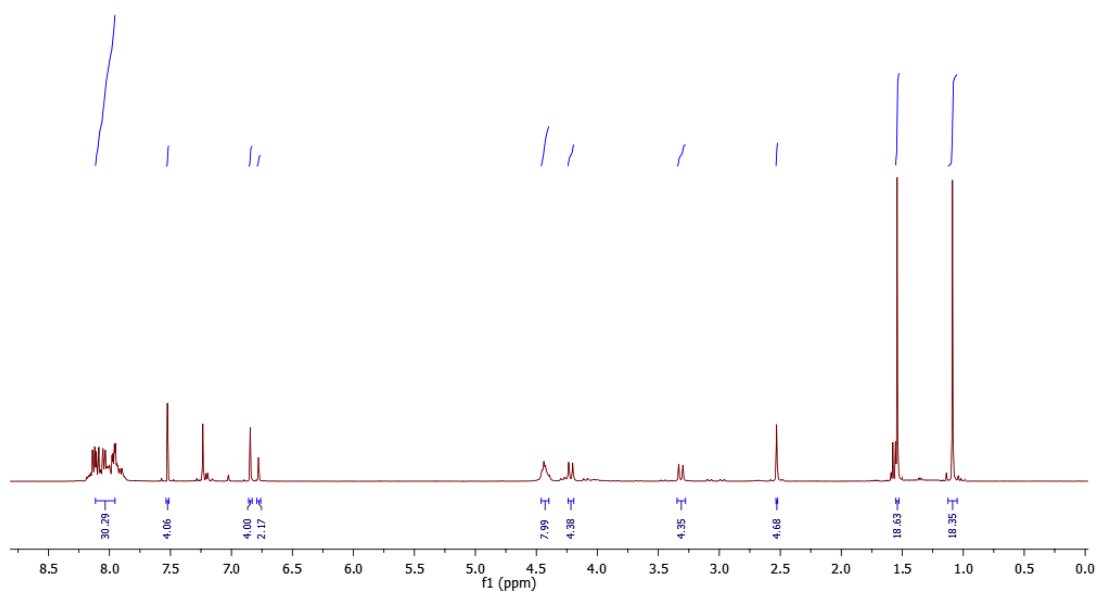


Figure 6. <sup>1</sup>H NMR spectrum of compound CP

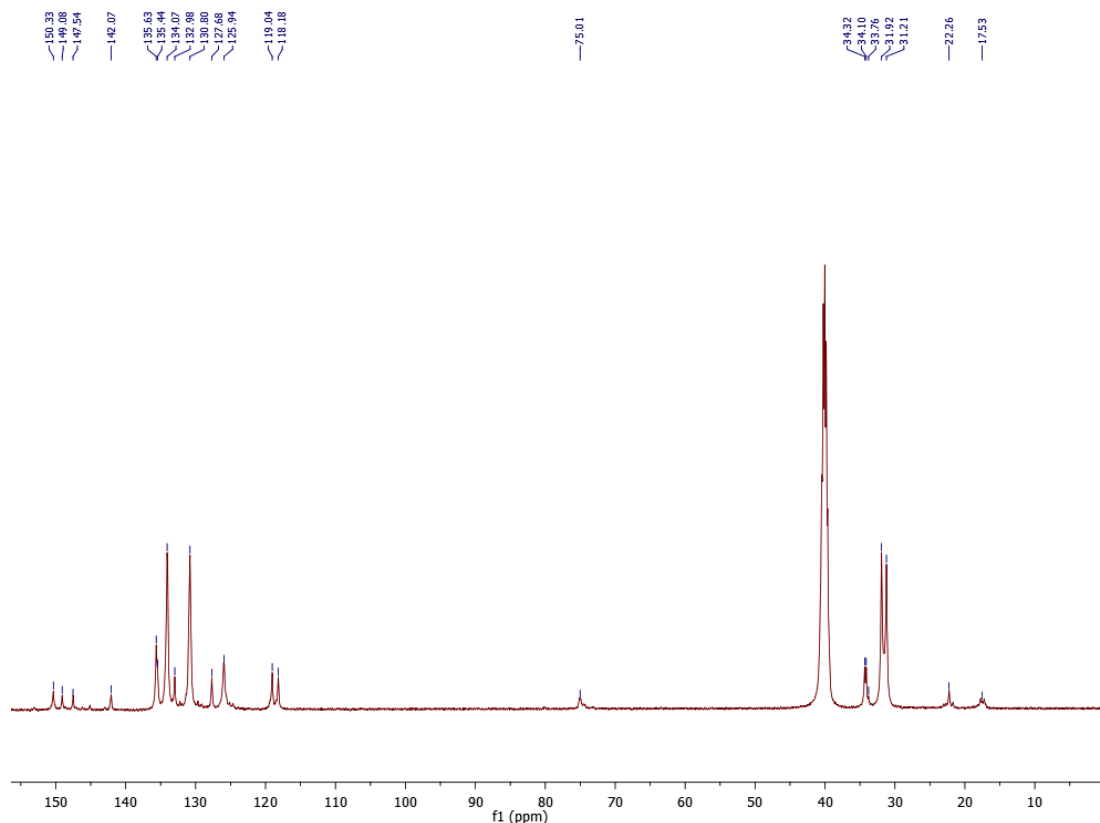


Figure 7.  $^{13}\text{C}$  NMR spectrum of compound CP

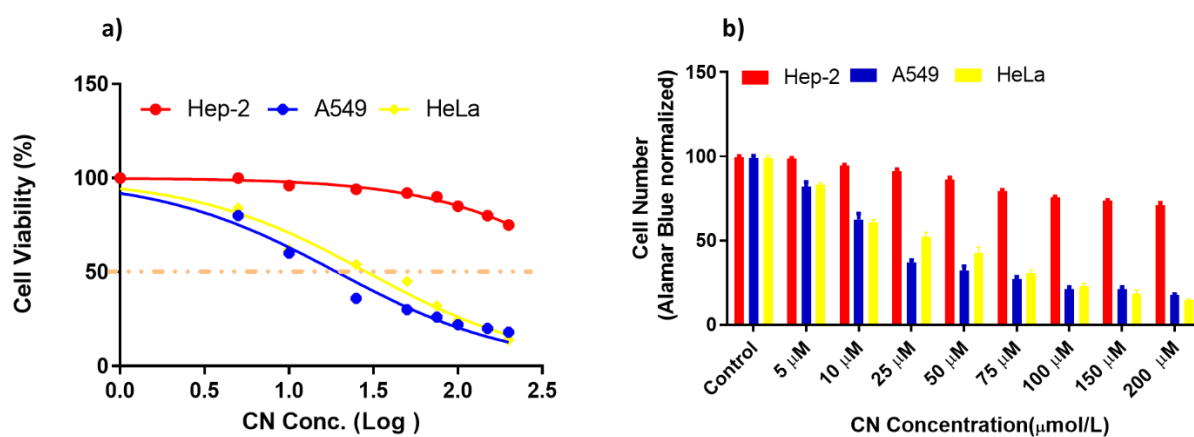
## 2.1. Cytotoxicity study of cationic calix[4]arene Derivatives (CN and CP)

In vitro cytotoxicity of CN and CP compounds was examined at various concentrations to determine their cytotoxic effects on the proliferation and viability of lung cancer cell line A549, cervix cancer cell line HeLa and epithelial cell line Hep-2 cells using the Alamar blue assay (Figure 8). CN and CP compounds inhibited the proliferation and viability of the lung and cervix cancer cell lines in a dose-dependent manner (Table 1). As shown in Table 1, compounds CN and CP were found to show high cytotoxic effect over human lung cancer cell (A549) and human cervix cancer cell (HeLa) and had  $\text{IC}_{50}$  values of  $13.42\ \mu\text{M}$  for CN and  $17.05\ \mu\text{M}$  for CP, respectively (Figure 9). In addition, CN and CP did not have cytotoxic effect on human epithelial cells (Hep-2) and their  $\text{IC}_{50}$  values were found to be  $162.71\ \mu\text{M}$  and  $>200\ \mu\text{M}$ , respectively.

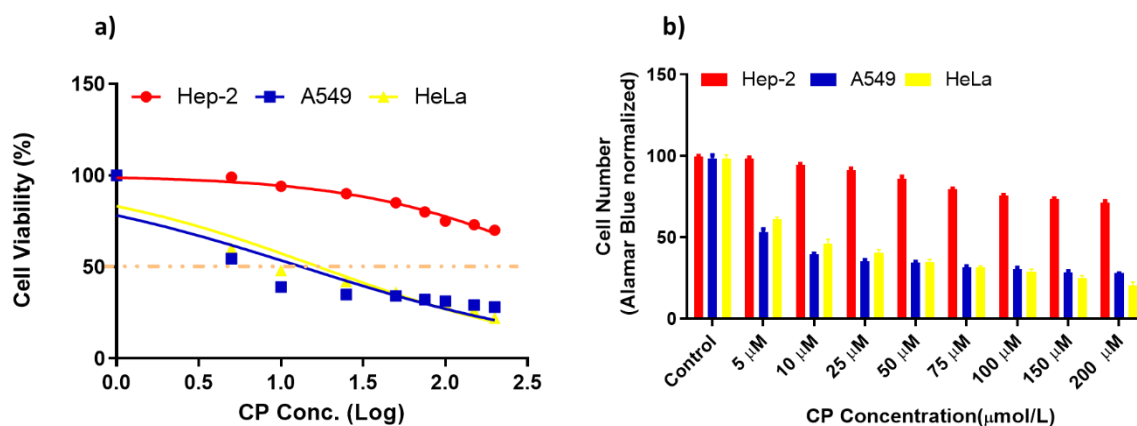
**Table 1.** IC<sub>50</sub> values (0-200 μM) of cationic calix[4]arenes on different cells

	Hep-2	A549	HeLa
CN	>200	13.42	28.96
CP	162.71	22.94	17.05
3a*	--	89	100
3d*	--	16	95

(\*) Ref. (An et al., 2016)



**Figure 8.** Influence of compound CN on the viability of A549, HeLa and Hep-2 cells. a) Alamar Blue test was utilized to analyze cytotoxicity. It was incubated for 48 hours with different concentrations in the range of 0-200 μM. b) IC<sub>50</sub> value of A549 and Hep-2 cells.



**Figure 9.** Influence of compound CN on the viability of A549, HeLa and Hep-2 cells. a) Alamar Blue test was utilized to analyze cytotoxicity. It was incubated for 48 hours with different concentrations in the range of 0-200 μM. b) IC<sub>50</sub> value of A549 and Hep-2 cells.



### 3. CONCLUSION

In summary, after successful synthesis of two cationic calixarene derivatives, their antitumor properties were evaluated. 5-Bromopentyl-trimethylammonium bromide derivative (CN) showed the highest toxicity against human A549 cell line (13.42  $\mu\text{M}$ ) and 3-bromopropyl-triphenyl-phosphonium bromide derivative (CP) showed the highest toxicity against human HeLa cell line (17.05  $\mu\text{M}$ ). Results show that CN and CP are desirable antiproliferative agents since no cytotoxic effects have been observed on healthy epithelium cells. Hope this study will find its applicability in the field of the drug industry.

### 4. CONFLICT OF INTEREST

The authors declare no conflict of interest.

### 5. ACKNOWLEDGMENTS

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