

The Environmental Importance of Flue Gas Purification Systems; Case of Yatağan Thermal Power Station

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ABSTRACT

During the production of energy from large combustion plants will cause the combustion process to air pollution stands out various gases. That gases are quite harmful on the urban ecosystem. Wet flue gas desulphurisation systems are preferred because of their applicability and easily operable structures without a deep information. The Station flue gas purification systems are designed as a gas desulphurisation process. Each of these units has their own chimney and fluepurification systems. In flue gas purification systems, SO₂, SO₃, dust and heavy metals are purified. SO₂ which occurs by fuel burning can be increased in flue gas purification systems with 98% efficiency. The process is worked on the basis of spraying the lime milk produced by grinding the limestone in the mill and mixing it with water, into the waste gas in the washing tower. The lime milk polluted after spraying process is segregated by way of hydro cyclones and discarded. The lime milk which diminishes is taken from the stock tank and given to the system. SO₂ gas which enters in the flue gas purification system can be increased to 500 mg/Nm³. It is seen how important the gas desulphurisation process in preventing environment pollution and for a healthy landscape. This study has been presented at the II. International Sustainable Buildings Symposium in the same form.

Keywords: Flue gas desulphurisation, SO₂ absorption, Yatağan thermal power plant, air quality, air pollution and environment.

Baca Gazı Arıtma Sistemlerinin Çevresel Önemi; Yatağan Termik Santrali Örneği

ÖZ

Büyük yakma tesislerinde enerji üretimi sırasında yakıtın yanması sonucu hava kirliliğine neden olacak çeşitli gazlar çıkmaktadır. Bu gazlar kent ekosistemi üzerinde oldukça zararlıdır. Yaş baca gazı desülfürizasyon sistemleri, uygulanabilirliği geniş ve kimyasal proseslerin yönetiminde derin bir bilgi gereksizinden dolayı tercih edilmektedir. Yatağan termik santrali baca gazı arıtma sistemleri yaş baca gazı desülfürizasyon prosesi olarak tasarlanmıştır. Ünitelerin her birinin kendi bacası ve kendi baca gazı arıtma sistemi vardır. Baca gazı arıtma sistemlerinde SO₂, SO₃, toz ve ağır metallerin arıtımı yapılmaktadır. Yakıtın yakılması sonucu ortaya çıkan SO₂ gazı, baca gazı arıtma sistemlerinde %98'e varan tutma verimiyle arıtılabilmektedir. Proses, belirli ölçülere kadar kırılmış kireç taşının değirmende öğütülmesi ve su ile karıştırılması sonucu hazırlanan kireç sütünün, yıkama kulesinde atık gazın üzerine püskürtülmesi esasına göre çalışmaktadır. Püskürtme sonrası kirlenen kireç sütü yoğunluk farkına göre hidro siklonlar vasıtasıyla ayrılmakta ve atılmaktadır. Eksik kalan kireç sütü ise belirli miktarda kireç sütü depolama tankından alınarak sisteme verilmektedir. Yatağan termik santrali örneğinde baca gazı arıtma sistemine giren SO₂ gazı 500 mg/Nm³'e kadar arıtılabilmektedir. Çıkan bu sonuçlar değerlendirildiğinde, santralin baca gazı desülfürizasyon proseslerinin çevre kirliliğini önlemesi ve sağlıklı bir peyzaj açısından ne kadar önemli olduğu ortaya çıkmaktadır. Bu çalışma II. Uluslararası Sürdürülebilir Yapılar Sempozyumu'nda aynı şekliyle sunulmuştur.

Anahtar Kelimeler: Baca gazı desülfürizasyon, SO₂ absorpsiyonu, Yatağan termik santrali, hava kalitesi, hava kirliliği ve çevre.

1. INTRODUCTION

In the larger burning stations, several gases occur which cause air pollution during the process of fuel burning.

These wastes cause air, water and soil pollution. This situation cause many illness and deterioration of the environmental landscape character. Plant facilities, water resources, agricultural lands, forested areas etc. of the near and far surrounding can be damaged. One of the main of this waste is SO₂. After burning the fuel, there becomes energy but after this burning process the gasses

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extracted to the air causes air pollution. Approximately 80% of SO₂ emissions are produced by fuel burning [1]. It is seen that after the catalysation by solid particles or sun rays or oxidation upon water drops by complex reactions, a minor part of SO₂ turns to SO₃ and after that it leaves atmosphere by solid dry deposit. It might be a fatal problem for the respiratory tract or lung failure patients when SO₂ in the air reaches higher degrees. Besides that it also harms the plants and technological materials. However, its most serious effect on nature is demolishing and destroying effect on the forests, flora, soil structure and water resources [2]. On the purpose of controlling and decelerating the air pollution in producing energy, gas purification systems are established in burning stations. Flue gas purification systems are purified the pollution in its resource and they minimise the damage in the air quality. A well-working flue gas purification station can decrease the harm level of the gases that occur after fuel burning process and produce clean energy in this way. On this ground, flue gas purification systems are crucial to control air pollution and to protect the air quality. The waste which occurs after coal burning can be discarded by applying either or few of physical, chemical or biological methods [3].

As a rule, the method of purification the SO₂, one of the contaminants found in flue gas, depends on the principle of transferring the contaminant to the purifier liquid prepared for the purpose of purification by reaction with flue gases. This transferring system depends on absorbing, ravelling or adsorption of the contaminant [4]. Flue gas purification systems purify the contaminants produced by coal burning in the thermal power stations in energy producing. These purified contaminants are SO₂, SO₃, dust and heavy metals. Flue gas purification systems are composed of flue gas washing tower, limestone stocking, the systems of preparing absorbent, cultivation, plaster dewatering and ash mixing and the other helper systems.

2. MATERIAL AND METHOD

Yatağan Termik Santralinde ıslak tip baca gazı desülfürizasyon sistemi kullanılır. Wet flue gas desulphurisation systems work as in the figure given below.

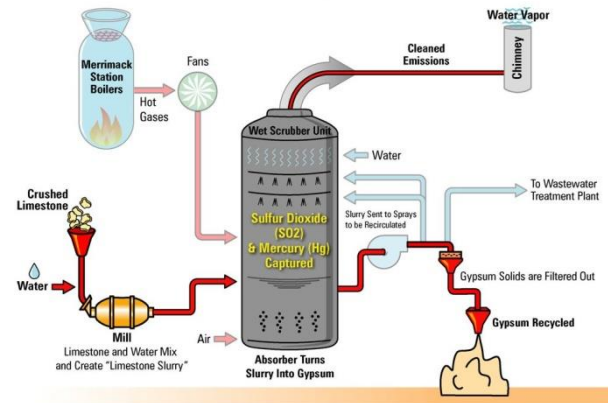
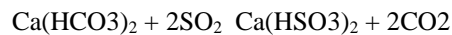


Figure 1. The desulphurisation process of wet chimney [5]

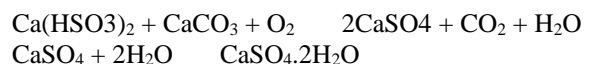
2.1. Lime Milk Preparation Process

The flue gas purification system of our case Yatağan Thermal Power Station is also composed of the systems given in the figure above. The limestone needed for preparing lime milk are stocked in the silos, measured in dimensions and sent to the crackers. The limestone cracked in specific dimensions are granulated and transformed into water and lime milk. Well-granulated limestone are dispersed in the water and it turns into a suspension [2]. Prepared lime milk is stocked in the absorbent tank and the needed amount of it is sent to the washing tower (scrubber) through the feeding bands. Using the recirculation pumps, lime milk is given to the distribution bands in the washing tower and then sprayed in such a way that it contact with raw gas with the approximate percentage of 100% by the help of nozzles. SO₂ found in flue gas is absorbed in the water and react with limestone.



2.2. Purification Process in the Washing Tower

While the processed gas is getting high, it passes through the drop catchers, dehumidified here and sent to the chimney from the fresh air exit. Calciumsulphide which occurs after the reaction of lime milk and the polluted gas is osculated to the oxygen and tuned into calcium sulphate. When the density in the was hing tower exceeds the treshhold limits (1130 kg/m³), the material is sent to the cyclones. Taking the advantage of the density gap, the one which has higher density is taken to the plaster tank and the other with lower density is taken again to the washing tower (scrubber). The reactions which occur in the washing tower through the purification process in the flue gas purification systems are given below.



The well-used wet and dry flue gas purification systems and the information of the products which occur after the reaction of the active materials in these systems are given below.

Table 1. The processes of flue gas desulphurisation mostly applied

Process	Active Material	Product
WET SYSTEMS		
Limestone	CaCO ₃ (discarded)	CaSO ₃ / CaSO ₄
Lime	Ca(OH) ₂ (discarded)	CaSO ₃ / CaSO ₄ (discarded)
Double alkaline	Na ₂ SO ₃ (regained)	CaSO ₃ / CaSO ₄
Sodium carbonate	Na ₂ CO ₃ /Na ₂ SO ₃ (Na ₂ CO ₃ /Na ₂ SO ₃ regained)	CaSO ₃ / CaSO ₄
Mag-Ox	MgO (regained)	SO ₂ (may be used in the production of S or H ₂ SO ₄)
Wellman-Lord	Na ₂ SO ₃ (regained)	H ₂ SO ₄
DRY SYSTEMS		
Limestone	CaCO ₃ (discarded)	CaSO ₄
Lime	Ca(OH) ₂ (discarded)	CaSO ₄
Trona	(Na ₂ CO ₃ - NaHCO ₃) (discarded)	Na ₂ SO ₄
Spray drying	Lime, lime stone, soda (trona)	DryCaSO ₃ orNaSO ₃

2.3. Chimney Gas Desulphurisation System Equipments

Our case Yatağan Thermal Plant is composed of the sub-systems below:

- Flue gas washing system
- Limestone stocking, operating and preparing absorbent system
- Dewatering plaster and ash jamming system
- Flue gas washing system of every unit is composed of the sub-systems below
- Flue gas way
- Scrubber
- Helper systems are composed of the sub-systems below
- Limestone system
- Limestone stocking, transporting and urgent stocking system
- Preparing and feeding absorbent
- Plaster dewatering
- Hydrocyclone station (Units 1-2-3)
- Gypsum parrying
- Filtrates recycle
- Transferring plaster mud

- Ash transfer
- Ash mixing
- Provision of process water
- Provision of the service and device air
- Drainage hole
- Urgent stocking tank
- Flue gas washing system
- Flue gas way: The elements of the way of flue gas which links the existing flue gas channels to scrubber are given below:
- Flue gas channels
- Booster fan
- Urgent spraying system
- Gas-gas warmer
- Air dryer
- The processes given below are done in washing tower
- Disposing SO₂
- Disposing the dust
- Turning the disposed SO₂ and O₂ to sulphuric acid
- Catching the drop
- Sub-equipments of the washing tower are given below:
- The entrance of flue gas
- Absorption region
- Drop catchers
- Oxidation region
- Crystallisation region

Scrubber pumps

The measurements of the analysis

Some of the technical information about the scrubber are given below

- Flue gas entrance section: 10 000×4200 mm
- Speed of the dirty flue gas exit: 12,7 m/s
- Absorption Region: 69 m³/h
- The speed of the gas entrance in drop catchers: 4,4 m/s
- The productivity of drop catcher: %99
- Number of washing tower: 3×1
- Type :Bischoff
- Gas flow rate (max) :1461595 Nm³/h (wet in the entrance)
- Tower diameter :13,5 m
- Height of tower :48,3 m
- Drop catchers :1 unit (two-level)
- Scrubber materials:carbon steel
- Spray level :6 level
- Number of mixers :4
- Material :carbon steel
- Nominal motor power:25 kW (each)
- Speed of the gas in scrubber :3,3 m/s

The height of oksidation region:from water level 5,4 m Grind ability : max 10 kWsaat/t
 Air temperature exiting from oxidation fan :75°C The capacity of reaction penetration : %50

Table 2. The attributions of constant measurement device

Place of measurement	Parameter	Model	Measurement range	Calibration Method
BGD Setting exit	CO	Siemens- Ultramat 23	0-750 mg/m ³	Automatic
	SO ₂		0-2.000 mg/m ³	Automatic
	NOX		0-2.000 mg/m ³	Automatic
	O ₂		0-%20	Automatic
	Dust	Sick AG- OMD41	0-450 mg/m ³	Automatic
BGD Setting entrance	Flow	Sensor PiotoPub	Method: Pressure Transmitter	

Oxidation fans

Total number:4 (one for reserve)
 Type :turbo
 Capacity:18 500 Nm³/hour
 Pressure:1,85 bar
 Height of crystallisationregion :13.6 m (from bottom to oxidation region)
 Circulation pumps: 3/scrubber
 Type :santrifüj
 Capacity:13 500 m³/h
 Pump pushing height :19,90 / 23,40 / 27,00 m
 Material :iron + rubber covering

Shaft;

Power:1 200 / 1330 / 1540 kW
 Speed :350-450 rpm
 Productivity :%87
 Salmastra type:mechanic
 Stimulation motor ;
 Number :3×3
 Power:1 250-1750 kW
 Voltage:6,3 kV
 Speed:1 500 rpm
 Measurements done
 There is pressure meter, density meter, pH meter and flow meter in the pump line
 The chemical attributions of limestone
 CaCO₃: min%90
 MgO: <%3
 SiO₂: max%4
 Moisture: %5

Particle size : 0-60 mm

3. PERSISTENT MEASUREMENT SYSTEMS IN FLUE GAS PURIFICATION SYSTEMS

SO₂, nitrogen oxide NOX (NO), carbon monoxide (CO), oxygen (O₂) and dust emission limits are persistently automatically monitored in the exit channels of the flue gas purification (FGP) systems. The attributions of the measurement devices are given in the table below.

3.1. Measurement Techniques in FGP Systems

Satisfying the isokinetic conditions and taking sample are important factors in doing right measurements. Taking isokinetic sample can be defined with taking the sample at an equal rate without any breakdown in flow conditions in the resource. The flow characteristics are crucially important in the samples which involve both of the gas and the particles. Taking a sample at any speed and point is sufficient if the flow speed and concentration are both stable in the sample taking point. But, in case of taking a sample from a resource which has unstable flow characteristic, expected concentration would be higher than the observed concentration when the speed of taking gas sample would be higher then the speed of the waste gas and vice versa. The characteristic of the flow speed are both valid for the gas and the particle materials [6].

In the table below isokinetic gas and dust samples are analysed within the scope of the principles that SKHKKY anticipates for the emission resource chimneys by using the standard measurement techniques given in the table. The way of isokinetic dust sampling is shown in the figure below.

Table 3. Standard measurement techniques using in emission measurements

Measured parameter	Standard method
Designation case points	EPA Method 1
Gas speed and flow measurement	EPA Method 2
Identifying the moisture context	EPA Method 4
Oxidation gases CO SO ₂ NO _x	EPA Method CTM 030 EPA Method 6C EPA Method CTM 022
Dust	EPA Method 17
Halogens	EPA 26

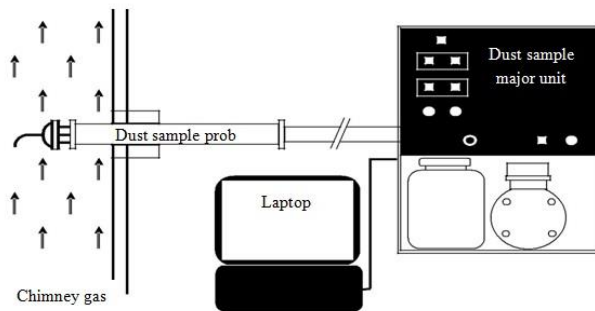


Figure 2. Isokinetic dust sample system

4. RESULTS OF THE MEASURES DONE IN FGP SYSTEMS AND THEIR EVALUATION

The emission measurement results of the dust, CO, NO_x and SO₂ parameters taken from flue gas measurement device for each three unit in the dates between January 2011 and December 2012 in CGP system are gained as daily averages.

4.1. Carbon Monoxide Emissions and the Evaluation

The results of the measures are shown in the graphics below where the CO emission values saved for each three unit are given with the threshold limits. From these graphics it is observed that CO emission data is rarely above from the threshold limits in terms of daily averages for Unit 1 and especially in Unit 3 its rate of being above extreme limit is higher. CO emission mostly increases parallel with the operation problems and negation in burning conditions. Looking these graphics it can be said that the burning conditions can be unfavourable.

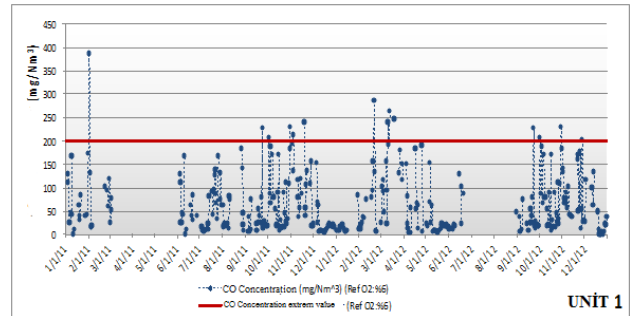


Figure 3. CO emission values and its comparison with threshold limit (Unit 1)

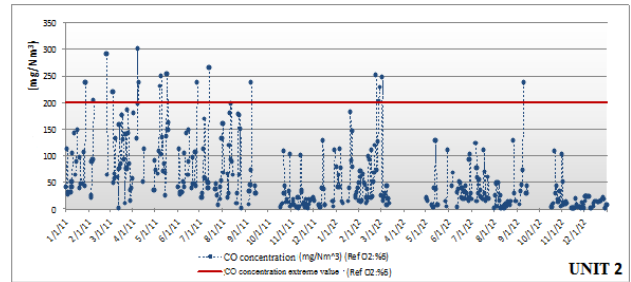


Figure 4. CO emission values and its comparison with threshold limit (Unit 2)

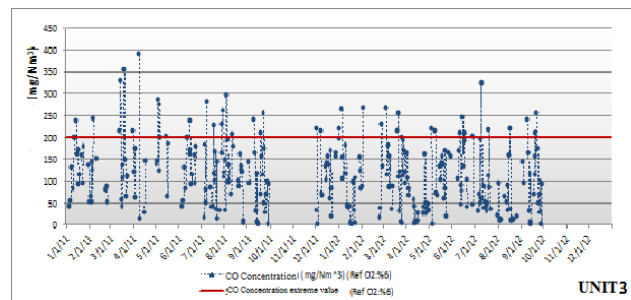


Figure 5. CO emission values and its comparison with threshold limit (Unit 3)

4.2. Sulphur Dioxide Emissions and the Evaluation

The results of the measurements are shown in the graphics below and the averages of the months where the number of measurements are below 10 are not calculated. In the graphics, SO₂ daily emission data saved in the units are given with the threshold limits. In terms of Daily averages, in each of these units it is observed that SO₂ concentrations is higher the threshold limit for some days, but in general, they are below from the threshold limit for each three units.

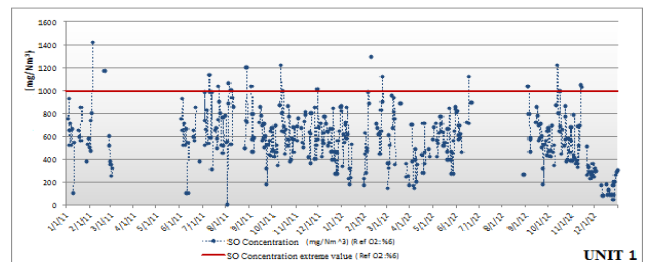


Figure 6. SO emission values and its comparison with threshold limit (Unit 1)

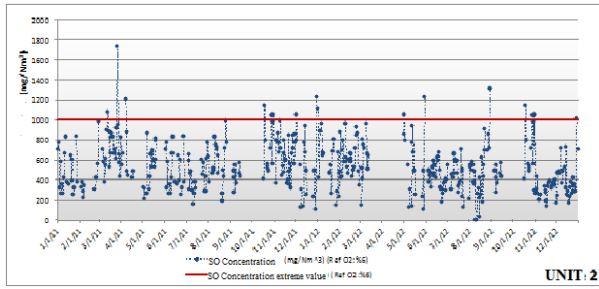


Figure 7. SO emission values and its comparison with threshold limit (Unit 2)

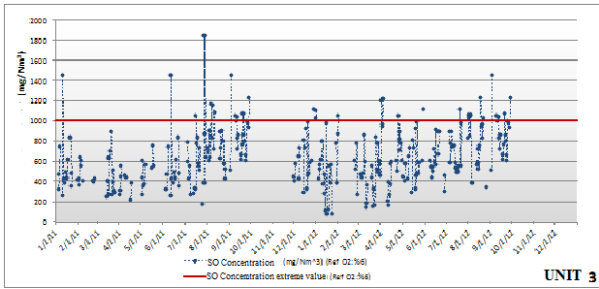


Figure 8. SO emission values and its comparison with threshold limit (Unit 3)

4.3. Nitrogen oxide Emissions and the Evaluation

The results of the measurements are shown in the graphics below and the averages of the months where the number of measurements are below 10 are not calculated. In the graphics, NOX daily emission data saved in the units are given with the threshold limits. As it is seen in the graphics, it not a matter of being higher than the threshold limit for three units in terms of daily emissions measured for the periods of January 2011 and December 2012.

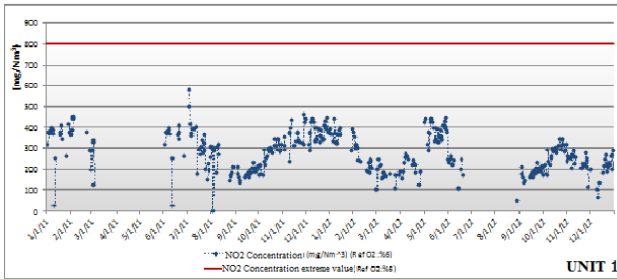


Figure 9. NOX emission values and its comparison with threshold limit (Unit 1)

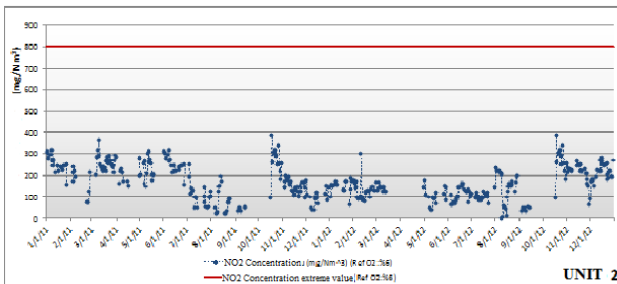


Figure 10. NOX emission values and its comparison with threshold limit (Unit 2)

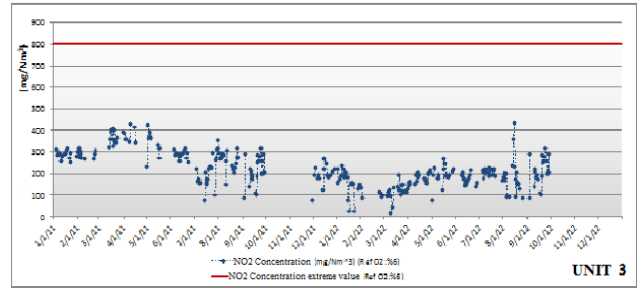


Figure 11. NOX emission values and its comparison with threshold limit (Unit 3)

4.4. Dust Emissions and the Evaluation

The results of the measurements are shown in the graphics below and daily averages are given with the threshold limits. In terms of daily averages, in every unit, there are very few days where the average values are higher than the threshold limits and in Unit 2 it is seen that emissions are closer to the threshold limits.

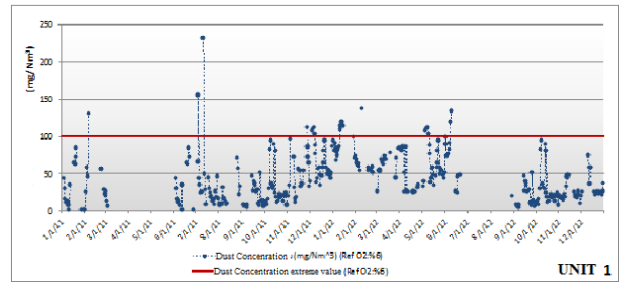


Figure 12. Dust emission values and its comparison with threshold limit (Unit 1)

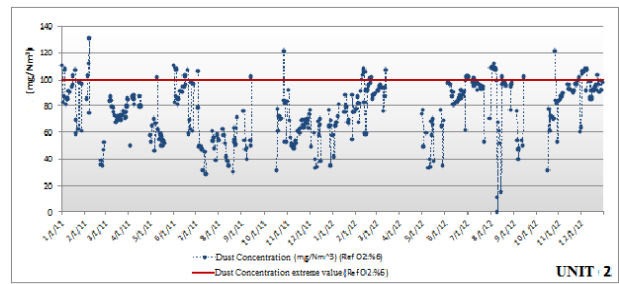


Figure 13. Dust emission values and its comparison with threshold limit (Unit 2)

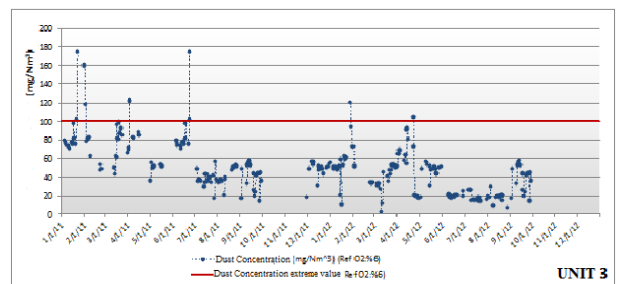


Figure 14. Dust emission values and its comparison with threshold limit (Unit 3)

5. CONCLUSION

When the daily average values gained from the station in the period of January-December 2012 are transformed into monthly averages, it is seen for every unit that the averages are not higher than the threshold limits for SO₂. In the station, threshold limits are reached for the three units. Nitrogen dioxide emissions gained from the flue gas purification system of the station are taken daily and the monthly averages are evaluated. In the daily measurements rises in the nitrogen dioxide emission are identified. In the station there exists a flue gas purification system. But, higher degrees in the emissions may be result of the flue gas purification station which does not work efficiently or the higher nitrogen values in the burning room. In order to hold the nitrogen dioxide emission below the threshold limit, it is required to be careful about preparing the lime milk that can occur in the operation conditions of the flue gas purification system and be able to ensure the station to work efficiently.

In the station, in every unit and for the mentioned period, the averages are not higher than the threshold limits for NO. Any rise is seen in the emissions of NO_x measured in the flue gas exit, but dust emissions are gained daily and the monthly averages are evaluated. In the dust emissions there are some levels where daily values are higher than the threshold limits. The electrostatic precipitator system of the station is renewed by annual revisions but it is seen that this renovation is now sufficient in some cases. In order to decrease the dust emissions, it is required to change or renew the electrostatic system.

In the station and for the period of January-December 2012, it is seen for every unit that the averages are not higher than the threshold limits for CO. For CO emission threshold limits are ensured. The data gained from the automatic analyser in the flue gas purification station are noted daily and the monthly averages are evaluated. In the CO emissions there are some levels where daily values are higher than the threshold limits. These higher degrees in the emissions may be the result of unfavourable burning conditions or the problems about the operation system. If the oxygen is getting lower and the fuel is getting higher in the burning room, CO emission would rise. In order to ensure CO not to be exceeded from the threshold limits, it is required to control the amount of the oxygen and fuel in the burning room and the unfavourable conditions which arise from the operation systems should be removed and, by this way, emission values should be stabilised at the threshold limit.

In this study it was concluded that Yatağan Thermal Power flue gas purification systems had not a negative effect on environmental health and landscape character. Yatağan Thermal Power does not create a permanent damage on the basic landscape parameters like air, water and soil quality.

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